# Stability of radicals in irradiated ammonia

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- Results for NH<sub>3</sub>
- Results for ND<sub>3</sub>
- Conclusions

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### Introduction

## Summary and Outlook

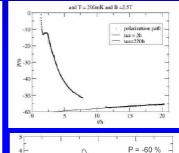
- High number of polarizable protons and deuterons
- Good resistance against radiation damage
- Radicals are stable in liquid nitrogen
- Relaxation time is comparable with older measurements
- It is still possible to polarize the ammonia to the known values
- The "old" SMC ammonia is still useable

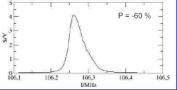
#### Outlook

- Temperature stability of the radicals

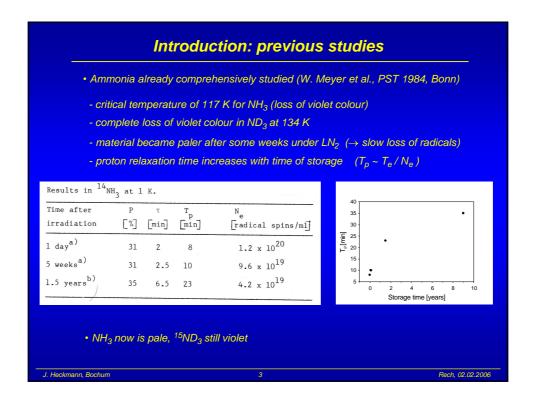
Miltenberg, 03.06.2005

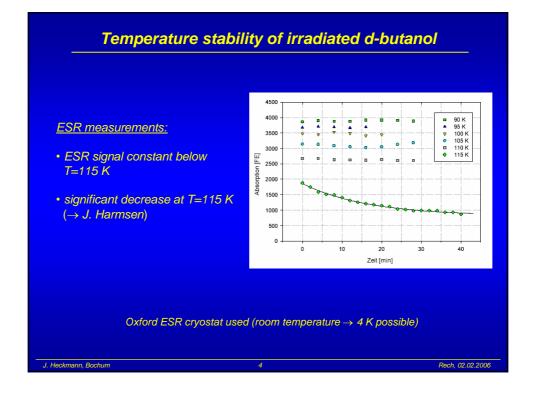
Daniel Buschert



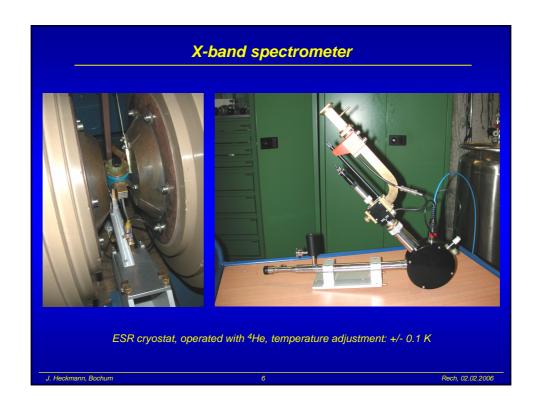


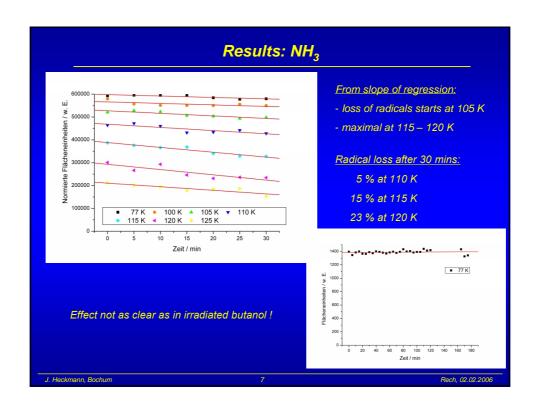
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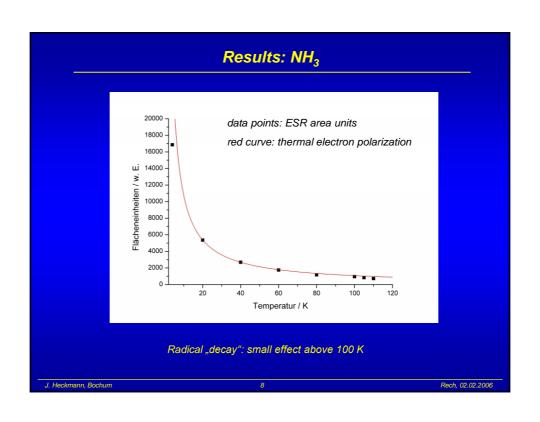


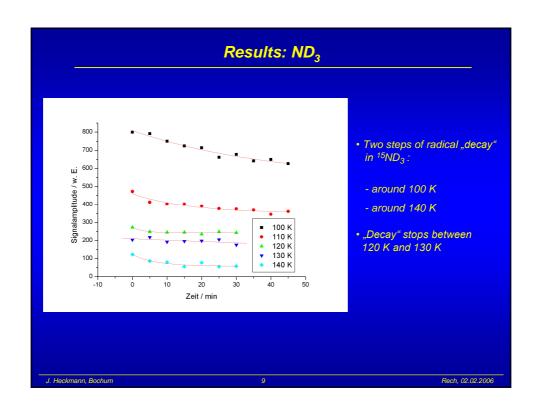


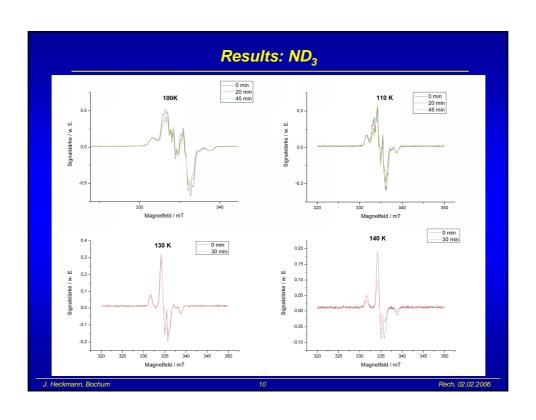












### **Conclusions**

- > No sharply defined critical temperature in NH<sub>3</sub>
  - radical "decay" starts at 105 K
  - maximal at 115 K 120 K (23 % of radicals lost after 30 minutes)
  - at 130 K all radicals are gone
  - colour center vanishes already at 77 K, DNP relevant center stable
- > Two steps of radical "decay" in ND<sub>3</sub>
  - first step around 100 K
  - second step around 140 K
  - ESR spectra: two different kinds of radicals
  - which center belongs to which "decay" step (analogous to NH<sub>3</sub>)?
  - $\rightarrow$  further studies (including <sup>14</sup>ND<sub>3</sub>)

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